

Er-Fe-V (Erbium-Iron-Vanadium)

V. Raghavan

Recently, [2007Kot] determined a partial isothermal section at 800 °C for Fe-rich alloys of this system, which depicts a ternary compound $\text{Er}(\text{Fe},\text{V})_{12}$.

Binary Systems

There are four line compounds in the Fe-Er system: $\text{Fe}_{17}\text{Er}_2$ ($\text{Ni}_{17}\text{Th}_2$ -type hexagonal), $\text{Fe}_{23}\text{Er}_6$ ($D8_a$, $\text{Mn}_{23}\text{Th}_6$ -type cubic), Fe_3Er (Ni_3Pu -type rhombohedral), and Fe_2Er ($C15$, MgCu_2 -type cubic) [Massalski2]. The Er-V phase diagram depicts a liquid miscibility gap, a monotectic reaction at 1870 °C, and a eutectic reaction near the Er-end at 1480 °C [Massalski2]. The Fe-V phase diagram [1984Smi] depicts one intermediate phase σ (31-66 at.% V; $D8_b$, σCrFe -type tetragonal).

Ternary Isothermal Section

With starting metals of 99.9% Er, 99.99% Fe and 99.99% V, [2007Kot] arc-melted alloys under Ar atm.

The samples were annealed at 800 °C for 500 h. The phase equilibria were studied with x-ray powder diffraction. The isothermal section constructed by [2007Kot] at 800 °C is shown in Fig. 1. The ternary compound $\text{ErFe}_{12-x}\text{V}_x$ ($1.6 \leq x \leq 2.3$, $D2_b$, ThMn_{12} -type tetragonal, $a = 0.84624\text{-}0.84660$ nm and $c = 0.47624\text{-}0.47648$ nm) is denoted τ here. The binary compound $\text{Fe}_{17}\text{Er}_2$ and Fe_2Er dissolve up to ~ 3 and ~ 8 at.% V respectively.

References

- 1984Smi: J.F. Smith, The Fe-V (Iron-Vanadium) System, *Bull. Alloy Phase Diagrams*, 1984, **5**(2), p 184-194
 2007Kot: B. Kotur, O. Myakush, and I. Zavalii, The Er-{Fe,Co}-{Ti,V} Systems and Hydrogenation Properties of the $\text{ErFe}_{2-x}\text{M}_x$ (M = Ti, V, Cr, Mn, Co, Ni, Cu, Mo) Alloys, *J. Alloys Compd.*, 2007, **442**, p 17-21

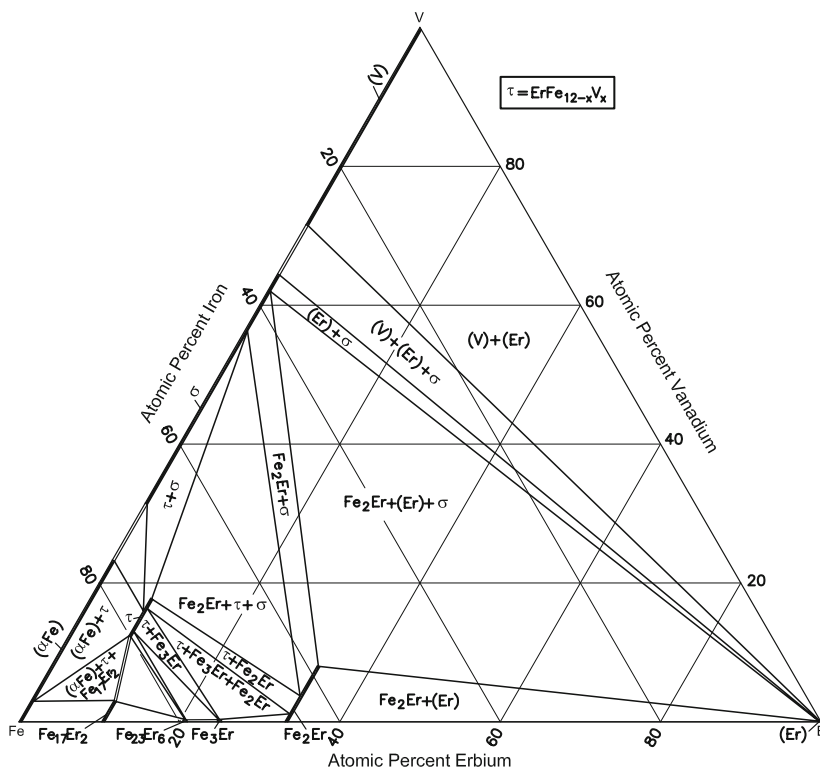


Fig. 1 Er-Fe-V partial isothermal section at 800 °C [2007Kot]